

Reproducing the Chemical Complexity of Sea Spray Aerosols in a Laboratory Setting

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1 **Supporting Information**

2 MART photobioreactor microcosm method

3 The Southern California Coastal Ocean Observing System (www.sccoos.org), a seawater
4 monitoring system off the coast of southern California, was used to monitor ocean conditions.
5 When ocean condition parameters such as salinity, sea surface temperature, and chlorophyll-a
6 concentration were within the location's monthly mean values, seawater was collected using a
7 cleaned bucket tied to a hoist at the end of Scripps Pier (275 m offshore) located in Scripps
8 Institution of Oceanography, University of California, San Diego. Seawater was filtered using 50
9 μm Nitex mesh prior to filling the MART. The 50 μm filtration removed any grazers that feed on
10 phytoplankton without altering other microbial species. Fresh seawater was added to a MART
11 system that had been previously cleaned with ethanol or isopropanol and triple rinsed with
12 deionized water.

13 Once the MART was filled, control measurements utilizing the freshly collected seawater
14 were performed, after which nutrients were added to the desired concentration. Another set of
15 control measurements were performed immediately after nutrient addition to account for the
16 change in seawater chemical composition due to the growth media. It was found empirically that
17 longer periods of aerosol generation led to insignificant phytoplankton growth, likely due to
18 phytoplankton cell damage in the centrifugal pump used to circulate water through the plunging
19 waterfall aerosol generation apparatus.¹ Phytoplankton growth was initiated by illuminating the
20 nutrient-doped seawater with two full spectrum fluorescent lamps (5700 K blackbody
21 temperature; Full Spectrum Solutions, 205457). During this initial growth period, water was
22 mixed and aerated by gently bubbling the sample by forcing particle-free air at 1 liter per minute

23 through 4-6 Tygon tubes (1/8 inch inside diameter) that were held on the bottom of the MART
24 by glass weights.

25 Once sufficient phytoplankton cell density was reached, determined empirically to be
26 approximately 12 mg chlorophyll m^{-3} , the bubbler system was removed and the MART was
27 sealed. The headspace of the MART was then purged with particle-free air. With the MART
28 purged of background particles, verified with a Condensation Particle Counter (TSI 3010), the
29 aerosol measurements were performed as described in the methods section. Aerosols were
30 generated with a two hours on, two hours off schedule. During aerosol generation, the waterfall
31 was operated with a 4 seconds on and 4 seconds off duty cycle to simulate the episodic nature of
32 natural breaking waves.¹⁻³ ATOFMS measurements were performed daily until at least one week
33 past the return of chlorophyll-a concentrations to that of the freshly collected seawater, in order
34 to capture chemical changes due to the biochemical processes associated with marine bacteria
35 and viruses. Each microcosm experiment lasted about 24 to 28 days total.

36 *In vivo* chlorophyll-a measurements were made at least once daily, and samples of the
37 bulk seawater for DOC, EEM measurements, and epifluorescence microscopy cell counts were
38 taken once daily. Samples for EEM analysis were analyzed the day of sampling. EEM excitation
39 and emission wavelengths ranged from 235-450 nm and 213-620 nm, respectively. EEM spectra
40 were blank subtracted using ultrapure water. Spectra were also corrected for inner-filter effects
41 and Rayleigh scatter masked (1st and 2nd order). To calibrate the EEM measurements, each
42 spectrum was normalized to the area of the water Raman scatter peak at 350 nm taken daily and
43 are reported in Raman Units (R.U.).⁴⁻⁵ Bulk seawater samples for epifluorescence microscopy
44 were taken daily, but SML and SSA particle samples for epifluorescence microscopy were taken
45 every two days. Fluorescence microscopy samples were pipetted into sterile cryogenic vials and

46 preserved with glutaraldehyde (0.05% electron microscopy grade). After an incubation period of
 47 15 minutes at approximately 4 °C, samples were flash frozen with liquid nitrogen and kept at -80
 48 °C until analysis.

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<i>Date</i>	<i>Chlorophyll-a (mg m⁻³)</i>	<i>Water Temp. (°C)</i>	<i>Pressure (dbar)</i>	<i>Salinity (PSU)</i>	<i>Tank Notation</i>
7/11/13 12:00	0.837	21.0916	4.132	33.6435	A
9/10/13 18:00	2.029	17.2773	5.542	33.4388	D
11/7/13 15:00	3.709	15.8678	5.171	33.4449	B
12/1/13 12:00	n/a	n/a	n/a	n/a	C
1/5/2014 16:30	1.346	15.0671	5.116	33.5286	E
4/11/201 4 12:00	4.491	14.3358	3.508	33.4264	F

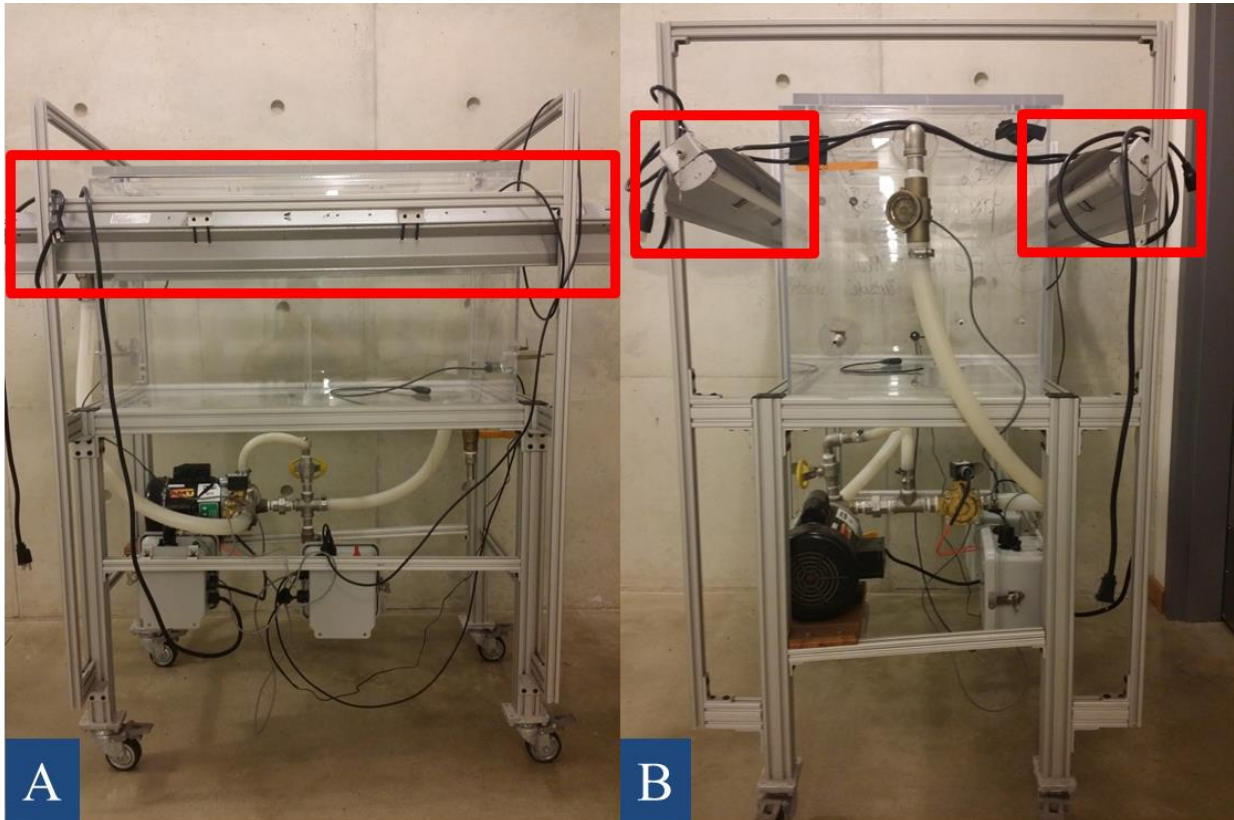
50 **Table S1.** Select metrics for the chemical conditions of the coastal Pacific Ocean at the time of
 51 seawater collection for each experiment. Data from SCCOOS were not available for the 12/1/13
 52 collection (Tank C).

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<i>Components</i>	<i>Molar Concentration</i>	
	<i>f/2 (M)</i>	<i>f/20 (M)</i>
NaNO ₃	8.82 x 10 ⁻⁴	8.82 x 10 ⁻⁵
NaH ₂ PO ₄ · H ₂ O	3.62 x 10 ⁻⁵	3.62 x 10 ⁻⁶
Na ₂ SiO ₃ · 9H ₂ O	1.06 x 10 ⁻⁴	1.06 x 10 ⁻⁵
FeCl ₃ · 6H ₂ O	1.17 x 10 ⁻⁵	1.17 x 10 ⁻⁶
Na ₂ EDTA · 2H ₂ O	1.17 x 10 ⁻⁵	1.17 x 10 ⁻⁶
CuSO ₄ · 5H ₂ O	3.93 x 10 ⁻⁸	3.93 x 10 ⁻⁹
Na ₂ MoO ₄ · 2H ₂ O	2.60 x 10 ⁻⁸	2.60 x 10 ⁻⁹
ZnSO ₄ · 7H ₂ O	7.65 x 10 ⁻⁸	7.65 x 10 ⁻⁹
CoCl ₂ · 6H ₂ O	4.20 x 10 ⁻⁸	4.20 x 10 ⁻⁹
MnCl ₂ · 4H ₂ O	9.10 x 10 ⁻⁷	9.10 x 10 ⁻⁸
Thiamine HCl (vit. B ₁)	2.96 x 10 ⁻⁷	2.96 x 10 ⁻⁸
Biotin (vit. H)	2.05 x 10 ⁻⁹	2.05 x 10 ⁻¹⁰
Cyanocobalamin (vit. B ₁₂)	3.69 x 10 ⁻¹⁰	3.69 x 10 ⁻¹¹

54 **Table S2.** Tabulated concentrations of nutrients in the final volume of seawater for higher
55 concentration (f/2) and lower concentration (f/20) nutrient additions. $\text{Na}_2\text{SiO}_3 \cdot 9\text{H}_2\text{O}$ is not part
56 of the ProLine nutrient mix, and was added separately.

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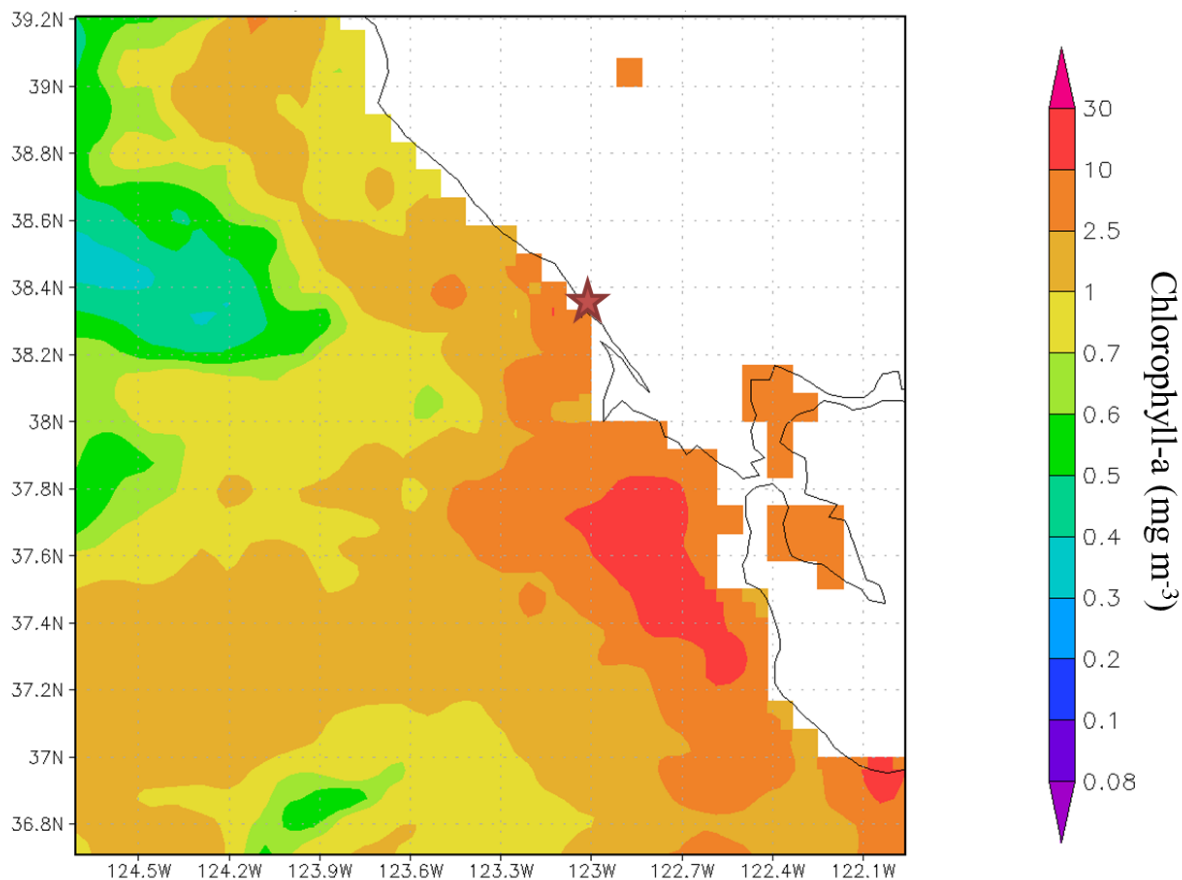


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59 **Figure S1.** Picture of modified MART photobioreactor. Highlighted boxes are two fluorescent
60 glow light fixtures that provide necessary illumination for growth of autotrophic microorganisms.
61 A. front view and B. side view.

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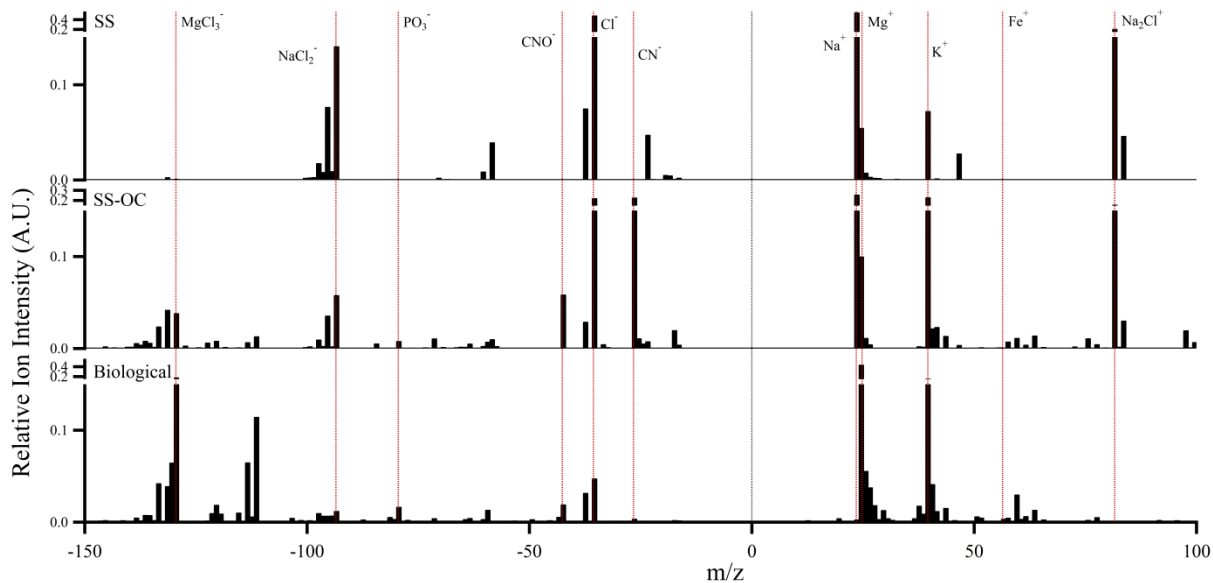
MODIS Chlorophyll-a Concentration
(06Mar2014 – 14Mar2014)



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64 **Figure S2.** Satellite-derived ocean surface chlorophyll-a concentration (MODIS) in the vicinity
65 of Bodega Bay, CA (red star). Chlorophyll-a concentration near the sampling location is $\sim 2 \text{ mg}$
66 m^{-3} . Wind direction and velocity measured at the time of sampling (313 ± 6 degrees $12.3 \pm 1.7 \text{ m}$
67 s^{-1}) suggest the air sampled is of primarily marine origin.

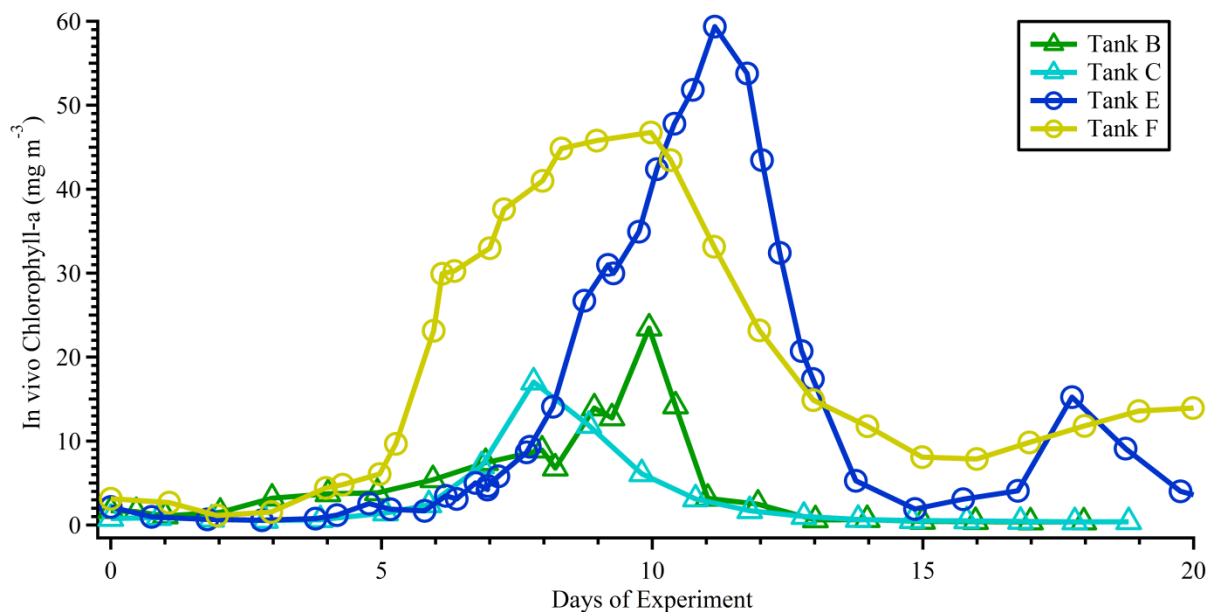
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70 **Figure S3.** Representative dual polarity mass spectra of 3 main particle types from ATOFMS
 71 observed in the microcosm experiments. From top to bottom panels, sea salt (SS), sea salt-
 72 organic carbon (SS-OC), and Biological type particle spectra are shown respectively.

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75 **Figure S4.** Compilation of absolute chlorophyll-a concentrations (mg m^{-3}) for four
 76 phytoplankton microcosms in this study (Tanks B, C, E, and F). Data from tanks A and D are not
 77 shown as the calibration for the chlorophyll-a concentration calculation during these two
 78 microcosms was not reliable. Initial chlorophyll-a concentrations for the two microcosms were
 79 provided by SCCOOS (Table S1).

80 **References**

- 81 1. Stokes, M. D.; Deane, G. B.; Prather, K.; Bertram, T. H.; Ruppel, M. J.; Ryder, O. S.;
82 Brady, J. M.; Zhao, D., A Marine Aerosol Reference Tank System as a Breaking Wave
83 Analogue for the Production of Foam and Sea-Spray Aerosols. *Atmospheric Measurement*
84 *Techniques* **2013**, *6*, 1085-1094.
- 85 2. Collins, D. B.; Zhao, D. F.; Ruppel, M. J.; Laskina, O.; Grandquist, J. R.; Modini, R. L.;
86 Stokes, M. D.; Russell, L. M.; Bertram, T. H.; Grassian, V. H. et al., Direct Aerosol Chemical
87 Composition Measurements to Evaluate the Physicochemical Differences between Controlled
88 Sea Spray Aerosol Generation Schemes. *Atmos. Meas. Tech.* **2014**, *7*, 3667-3683.
- 89 3. Callaghan, A. H.; Deane, G. B.; Stokes, M. D.; Ward, B., Observed Variation in the
90 Decay Time of Oceanic Whitecap Foam. *Journal of Geophysical Research-Oceans* **2012**, *117*.
- 91 4. Lawaetz, A. J.; Stedmon, C. A., Fluorescence Intensity Calibration Using the Raman
92 Scatter Peak of Water. *Applied Spectroscopy* **2009**, *63*, 936-940.
- 93 5. Murphy, K. R., A Note on Determining the Extent of the Water Raman Peak in
94 Fluorescence Spectroscopy. *Applied Spectroscopy* **2011**, *65*, 233-236.

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